## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

**International General Certificate of Secondary Education** 

## MARK SCHEME for the May/June 2011 question paper for the guidance of teachers

## 0654 CO-ORDINATED SCIENCES

0654/23

Paper 2 (Core Theory), maximum raw mark 100

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2011 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

	Page 2		Mark Scheme. Teachers Version	Syllabus	Faper
			IGCSE – May/June 2011	0654	23
1	(a) (i)	time actio	rence to: scale / time to renew ; on of heat / pressure ; on of microorganisms / decay ;		[max 2
	(ii)	oxyg	gen;		[1]
	(iii)		ose molecules join / link together ; rm long chains ;		[2
	<b>(b)</b> (C <sub>6</sub>	H <sub>14</sub> ) la	argest / heaviest ;		[1
	(c) (i)		gen ; er (vapour) ;		[2]
	(ii)		gas with) limewater ; s cloudy ;		[2]
	(iii)		on monoxide ; gen dioxide ;		[2]
					[Total: 12]
•		000/	energy/time; 600 = 13.3;		[3]
	(b) (i)		= $\frac{1}{2}$ mv <sup>2</sup> ; 5 × 2 × 40 × 40 = 1600 (J);		[2]
	(ii)		J (or same answer as <b>(i)</b> ) ; rgy is conserved ;		[2]

Mark Scheme: Teachers' version

**Syllabus** 

**Paper** 

Page 2

(c) expanded polystyrene / air / gas is a poor conductor of heat;

trapped air cannot carry heat around by convection;

concrete block is a poor conductor of heat;

aluminium reflects heat back into house;

[max 3]

[Total: 10]

	Page 3		3	Mark Scheme: Teachers' version	Syllabus	Paper
				IGCSE – May/June 2011	0654	23
3	(a)		A: tra			[2]
	(b)		stea from	eases; dy/linear, (increase); 0.6 to 1.1 (g/cm³)/by 0.5 (g/cm³);		[max 2]
		(ii)		e foods contain calcium ; ded for bones ;		[2]
		(iii)	any	citrus fruit / blackcurrants / other valid food source;		[1]
	(c)	``	`	e is) harder/stronger/less elastic/less smooth;		[1]
		(ii)		the surface of the bones) at the joint; ices friction / allows bones to move smoothly over ea	ach other;	[2]
						[Total: 13]
4	(a)			ne = force × distance ; 55 = 38500 (J) ;		[2]
	(b)	(i)	50 s	;		[1]
		(ii)		stant speed ; 5 m/s ;		[2]
	(c)	poi	nted e	hip between pressure, force and area ; end has small area and large pressure ; large area and small pressure ;		[max 2]
	(d)	less	s fricti	on;		[1]
						[Total: 8]

Page 4		,	Mark Scheme: Teachers' version	Syllabus	Paper	
				IGCSE – May/June 2011	0654	23
5	(a)	(i)	hair	[1]		
		(ii)	large	; [1]		
	(b)	(i)	diffusion ; from alveoli ;			
		(ii)	more oxygen can be absorbed (from the air)/taken in by lungs/compensates for less oxygen; more oxygen supplied to cells;			
			for respiration;			[max 2]
	(c)	(i)	ref. to limiting factors ; not enough grass to eat ; many eaten by, foxes / pumas ;			
		(ii)	idea won'	to species diversity; of their importance in food chain/provide food food to become extinct;	or pumas/so pui	
			otne	er, e.g. tourism / moral arguments ;		[max 2]
						[Total: 10]
6	(a)	(i)	Grou	up <b>1</b> , Period <b>2</b> ;		[1]
		(ii)	lithium is (very) reactive / easily combines with other elements / substances oil prevents reaction with air / oxygen / water / forms a protective barrier;			s; [2]
		(iii) lithium atoms have two shells / only have two electrons in first shell; lithium atoms have three electrons;				[2]
	(b)	(i)	hydr	rochloric (acid) ;		[1]
		(ii)	carb	oon dioxide ;		[1]
		(iii)	chlo	rine;		[1]
	(c)	(i)	subs	stance which changes the way the body works ;		[1]
		(ii)	,			
			avoid harming the user ; ensure correct dosage / owtte ;			[max 1]
					[Total: 10]	

	Page 5	5	Mark Scheme: Teachers' version Syllal		Paper
			IGCSE – May/June 2011	0654	23
7	(a) stra		es ; gles of incidence and reflection (correct by eye) ;		[2]
	<b>(b)</b> cor	rect dia	agram ;		[1]
	(c) (i)	red, bl	lue, green ;		[1]
	(ii)	freque	ency <b>or</b> wavelength ;		[max 1]
					[Total: 5]
8	(a) (i)	petals	s / nectary ;		[1]
	(ii)	anther	r / stamen ;		[1]
	(iii)	ovule	·,		[1]
	(fer	tilisatio	n is) the transfer of pollen from anther to stigma; n is) the fusion of male and female gametes; takes place before fertilisation;		[max 2]
	(c) (i)	17;			[1]
	(ii)	nucleu	us;		[1]
	(iii)	DNA ;			[1]
	(d) (i)	transp as suc	s produced by photosynthesis in leaves ; ported to flowers in phloem ; crose ; al ions in xylem ;		[max 2]
	(ii)	for res	spiration / for energy / to make nectar / any energy-re	equiring process;	[1]
					[Total: 11]

	Page 6		<b>i</b>	Mark Scheme: Teachers' version	Syllabus	Paper
				IGCSE – May/June 2011	0654	23
9	(a)	(i)	acid use surfa	perature ; concentration ; the same acid ; ace area of the metal ; me of acid ;		[max 3]
		(ii)	) ignites / pops ; hydrogen is given off ;			
	` '			A and C did not react/cannot decide between als did not react;	A and C/two of	the [1]
	(b)	(b) (i) electrolyte in beaker ; electrodes in electrolyte ; voltmeter connecting electrodes ;				
		(ii)		nge changes ; nuse voltage depends on the metals used for electro	odes ;	[2]
						[Total: 11]
10	(a)	(i)	uran	ium ;		[1]
		(ii)	nucle ener turbi			[3]
	(b)	(i)	lead	or concrete ;		[1]
		(ii)		ses ionisation inside cells ; ages cells / kills cells / mutation / damages DNA ; ser ;		[2]
			radia	ation sickness ; ation burns / burns skin ;		[max 2]
	(c)	(i)	Geig	er counter / GM tube etc. ;		[1]
	` '			lf-lives ; (years) ;		[2]
						[Total: 10]