Specimen Paper

Centre Number				Candidate Number				For Exam	iner's Use
Surname									
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Candidate Signature								Examine	s muais
		0	o 1'		_			Question	Mark



General Certificate of Secondary Education Foundation Tier

Additional Science 2 Unit 6

For this paper you must have:

- a ruler
- the Chemistry Data Sheet (enclosed)
- the Physics Equation Sheet (enclosed).

You may use a calculator.

Time allowed

• 90 minutes

Instructions

- Use black ink or black ball-point pen.
- Fill in the boxes at the top of this page.
- Answer all questions.
- You must answer the questions in the space provided. Do not write outside the box around each page or on blank pages.
- Do all rough work in this book. Cross through any work you do not want to be marked.

Information

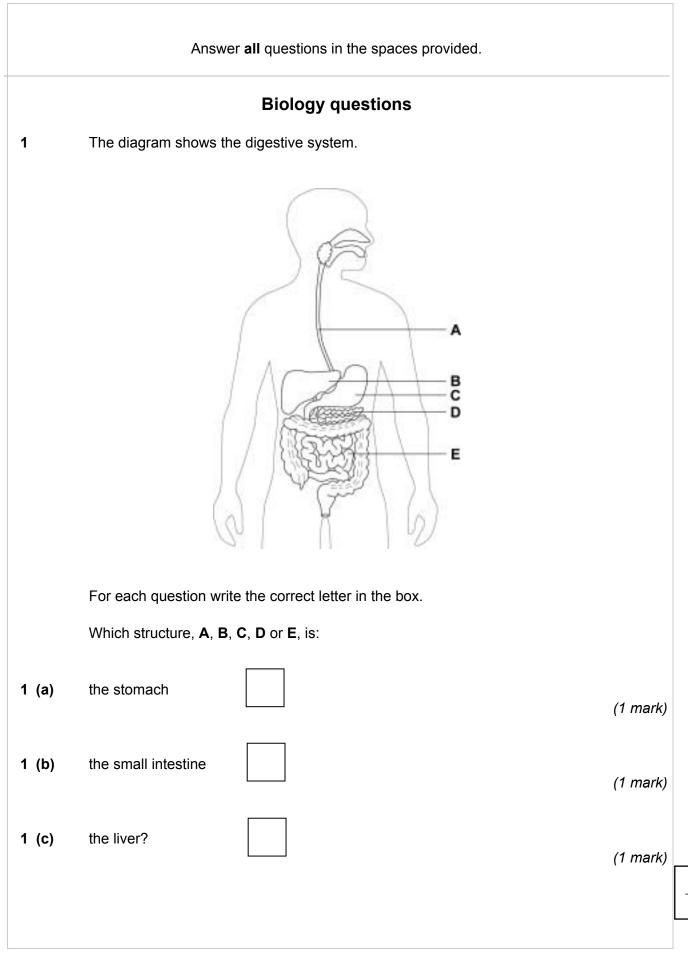
- The marks for questions are shown in brackets.
- The maximum mark for this paper is 90.
- You are expected to use a calculator where appropriate.
- You are reminded of the need for good English and clear presentation in your answers.
- Question 9 should be answered in continuous prose. In this question you will be marked on your ability to:
 - -use good English
 - -organise information clearly
 - -use specialist vocabulary where appropriate.

Advice

• In all calculations, show clearly how you work out your answer.

Examine	r's Initials
Question	Mark
1	
2	
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11	
12	
13	
14	
15	
TOTAL	







(2 marks)

Babies find it difficult to digest proteins in their food.

2

Baby food manufacturers use enzymes to 'pre-digest' the protein in baby food. Use words from the box to complete the following sentences. 2 (a) amino acids amylases fatty acids proteases glucose lipases Proteins are 'pre-digested' using enzymes called The pre-digestion of protein produces

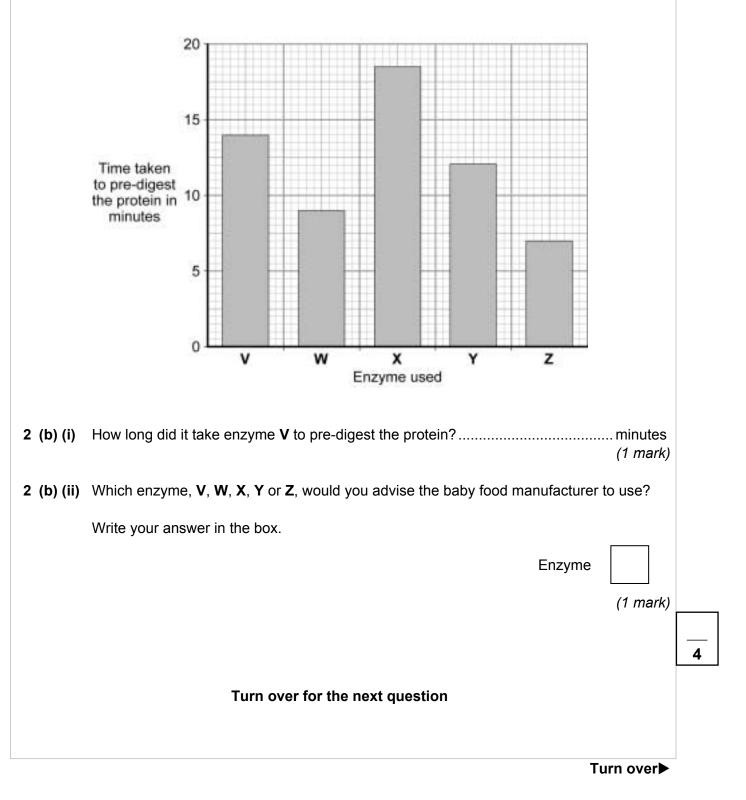
GCSE Additional Science 2 Unit 6F Specimen Paper Foundation Tier V1.0

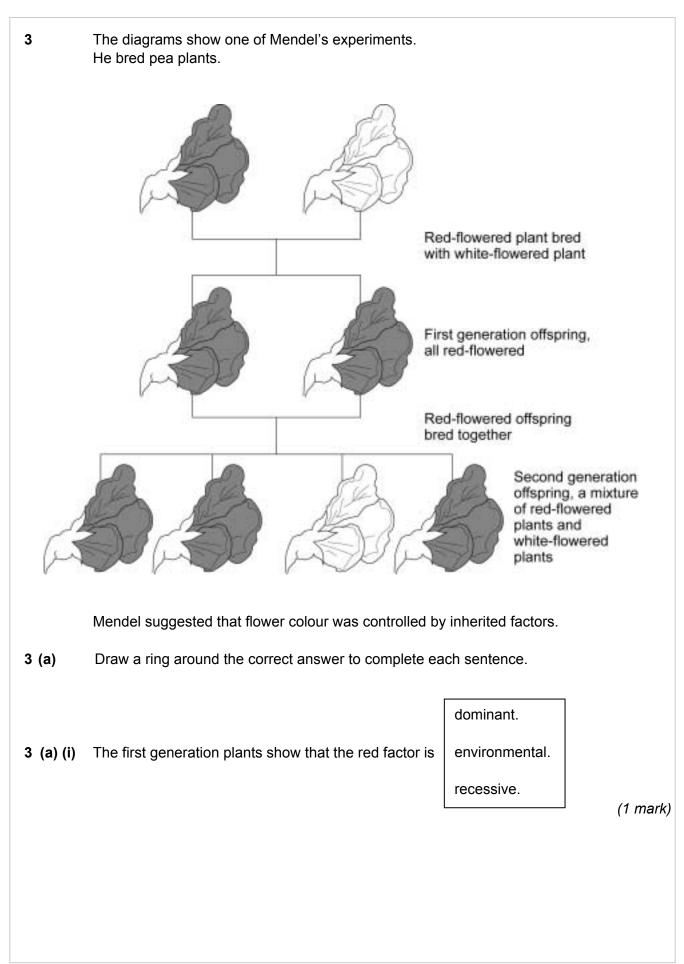
2 (b) A baby food manufacturer uses enzyme **V** to pre-digest protein.

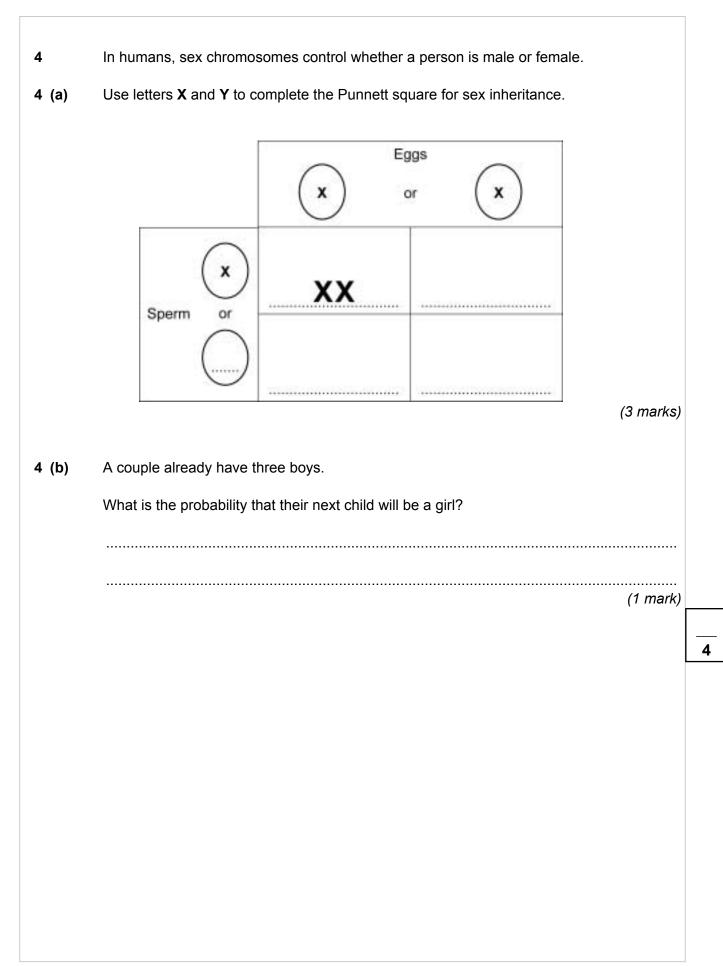
He tries four new enzymes, **W**, **X**, **Y** and **Z**, to see if he can reduce the time taken to pre-digest the protein.

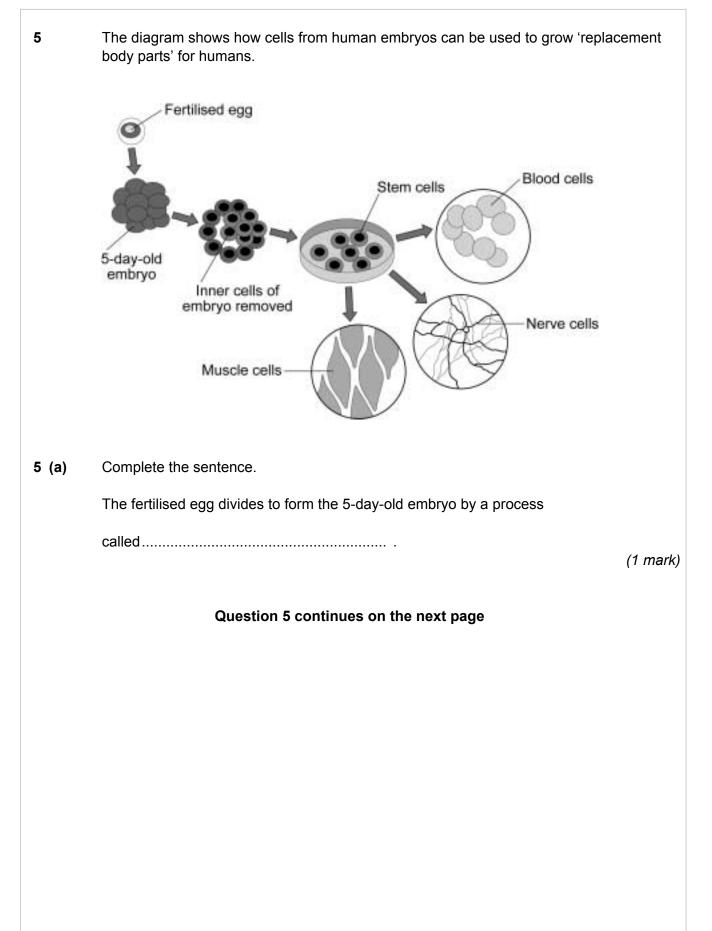
The graph shows the time taken for the enzymes to completely pre-digest the protein.

The manufacturer uses the same concentration of enzyme and the same mass of protein in each experiment.









5 (b) Some statements about stem cells are given below.

Tick (\checkmark) **two** advantages and **two** disadvantages of using stem cells to grow 'replacement body parts' for humans.

	Advantage Tick (√)	Disadvantage Tick (√)
Stem cells can grow into many different kinds of body cells.		
Stem cells may grow out of control.		
Large numbers of stem cells can be grown in the laboratory.		
Stem cells may be used to treat some human diseases.		
Collecting and growing stem cells is expensive.		
Patients treated with stem cells may need to take drugs for the rest of their life to prevent rejection.		

(4 marks)

	Chemistry Questions	
6	Ammonia has the formula NH_{3} . It is made from nitrogen and hydrogen. Ammonia dissolves in water to form a solution with a pH of 10.	
6 (a)	What does this pH value tell you about ammonia solution?	
	(1 m	 ark)
6 (b)	In industry a large amount of ammonia is neutralised by an acid to make ammonium nitrate.	
6 (b) (i)	What type of substance is ammonium nitrate?	
	(1 m	ark)
6 (b) (ii)	Which acid is added to ammonia to make ammonium nitrate?	
	(1 m	 ark)
6 (b) (iii)	Draw a ring around the main use of ammonium nitrate.	
	fertiliser lubricating oil medicine plastic (1 ma	ark)
	Question 6 continues on the next page	

6 (c) Instant cold packs are used to treat sports injuries.



One type of cold pack has a plastic bag containing water. Inside the bag is a smaller bag containing ammonium nitrate.

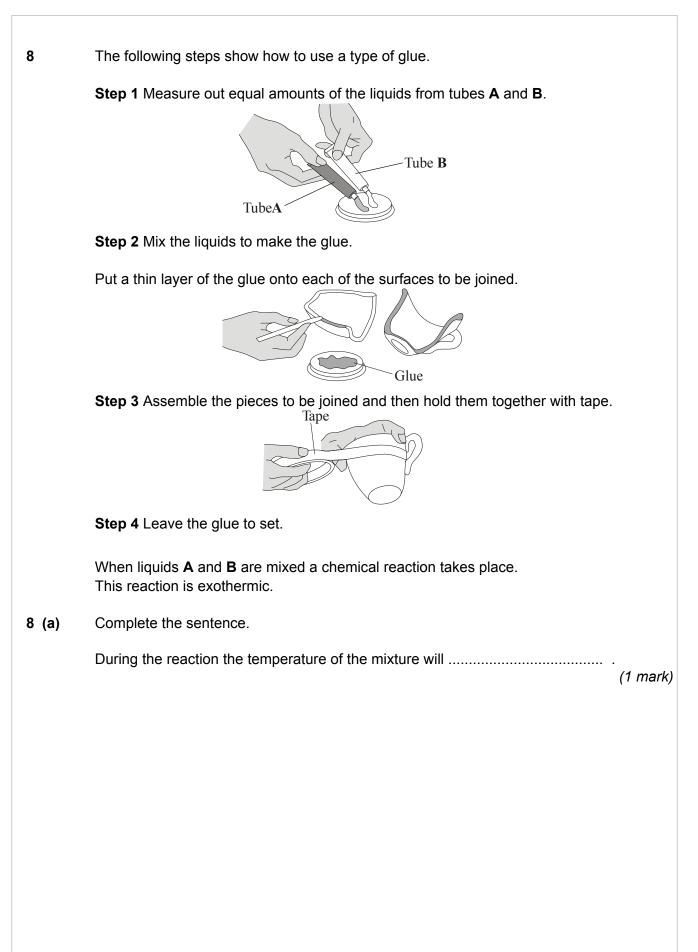
The outer bag is squeezed so that the inner bag bursts. The ammonium nitrate dissolves in the water. This process is endothermic.

Explain why the bag becomes cold.

.....

(2 marks)

7	Waste water from some industrial processes contains metal ions, such as chromium ions. These ions must be removed from the water before the water is returned to a river.		
	The equation shows a method of removing chromium ions from water.		
	CrCl₃(aq) + 3NaOH(aq) — → Cr(OH)₃(s) + 3NaCl(aq)		
7 (a)	This type of reaction is called a precipitation reaction.		
	What happens in a precipitation reaction?		
7 (b) (i)	Complete the name of the substance with the formula NaOH.		
	Sodium(1 mark)		
7 (b) (ii)	How many atoms are in the formula for NaOH?		
7 (c)	It is important to remove chromium ions from water before it is returned to a river.		
	Suggest why.		
	(1 mark)		
	4		



8 (b) The time taken for the glue to set at different temperatures is given in the table below.

Temperature in °C	Time taken for the glue to set
20	3 days
60	6 hours
90	1 hour

8 (b) (i) Complete the sentences below using words or phrases from the box.

decreases increases stays the same

When the temperature is increased the time taken for the glue to set

When the temperature is increased the rate of the setting reaction

(2 marks)

8 (b) (ii) Put a tick (✓) next to the **two** reasons why an increase in temperature affects the rate of reaction.

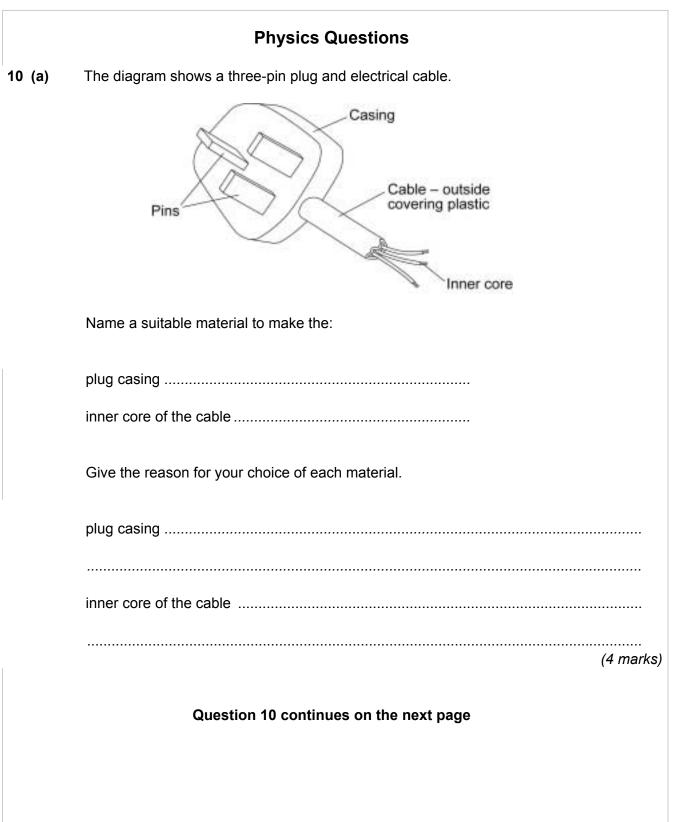
Reason	(~)
It gives the particles more energy.	
It increases the concentration of the particles.	
It increases the surface area of the particles.	
It makes the particles move faster.	

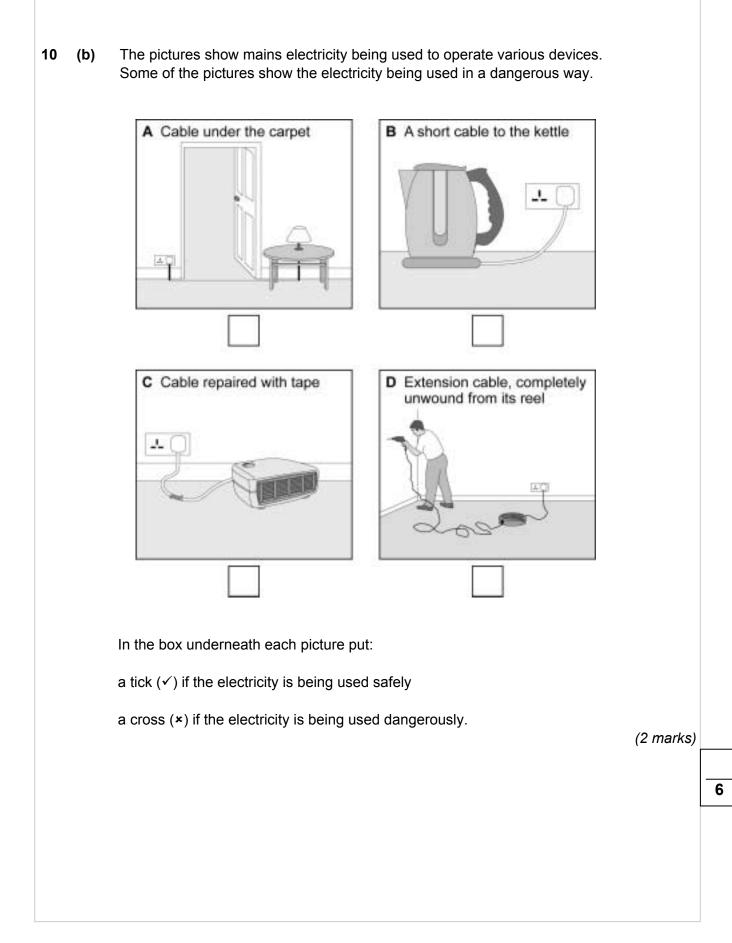
(2 marks)

5

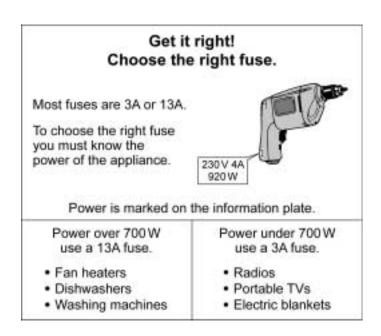
Turn over for the next question

9 In this question you will be assessed on using good English, organising information clearly and using specialist terms where appropriate. Copper sulfate crystals can be made from copper oxide and dilute sulfuric acid. Dilute sulfuric Copper oxide acid 1.04 Describe a method to make copper sulfate crystals from copper oxide and dilute sulfuric acid. For the method you should include: • the names of the pieces of apparatus used • a risk assessment. (6 marks)





11 (a) Look at the electrical safety information poster.



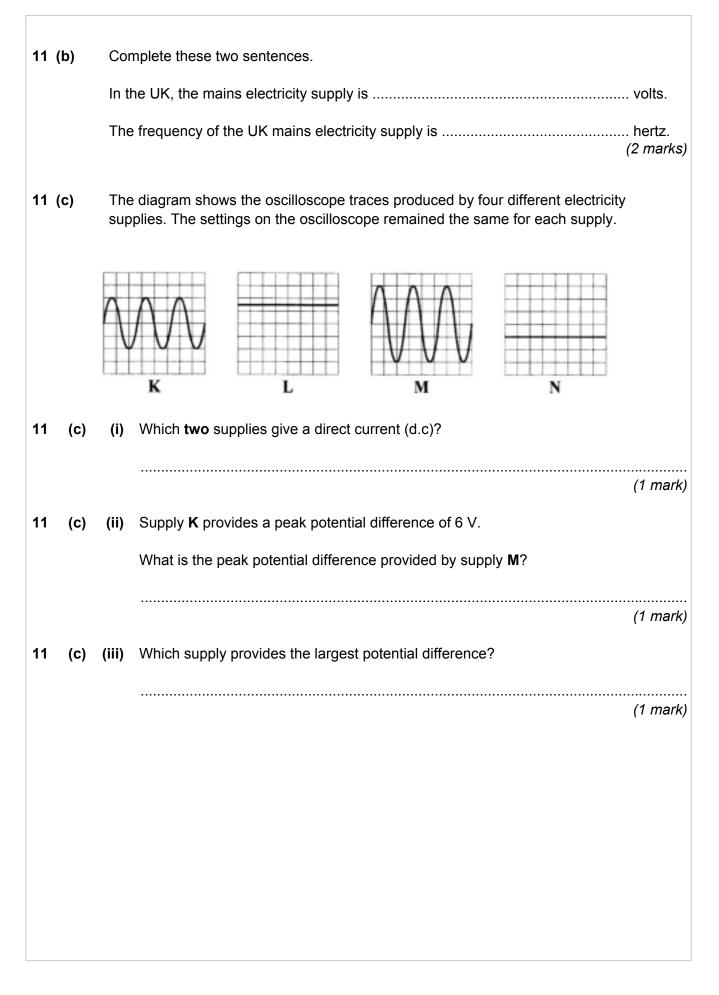
11 (a) (i) Complete the table to show which size fuse, 3 A or 13 A, should be fitted to each of the appliances.

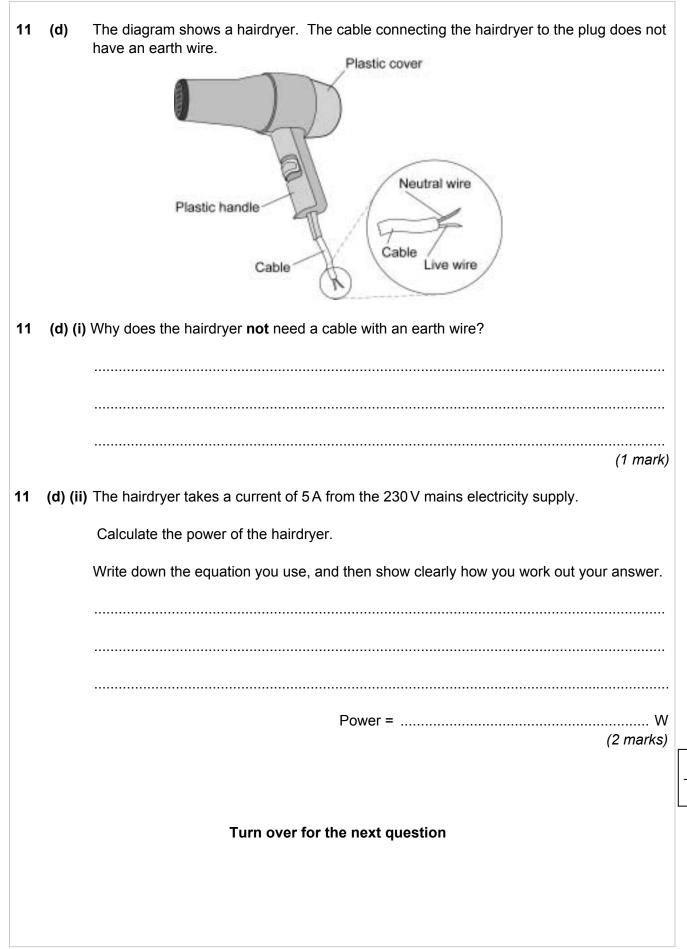
Appliance	Power	Fuse
Kettle	2200 W	
Hair straighteners	75 W	
Coffee maker	1260 W	

(2 marks)

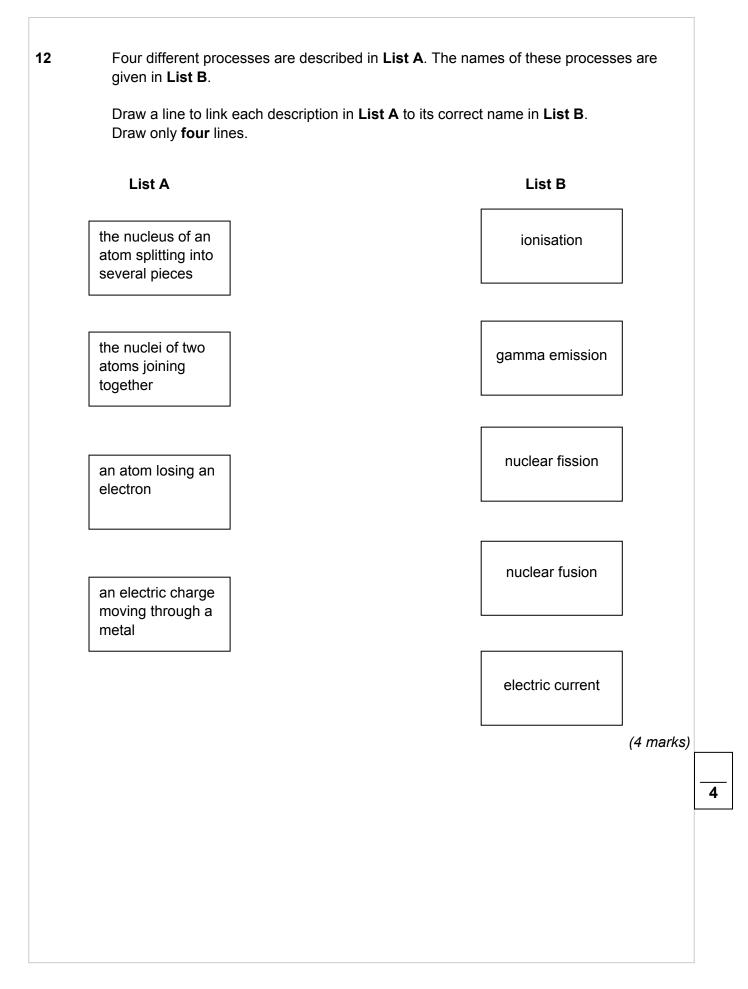
11 (a) (ii) The plug of a washing machine has been wrongly fitted with a 3A fuse.

Explain why the washing machine stops working shortly after it is switched on.





Turn over▶



Biology questions

13	Energy is obtained from both aerobic and anaerobic respiration during exercise.
13 (a)	Give three differences between aerobic and anaerobic respiration.
	1
	2
	3
	(3 marks)

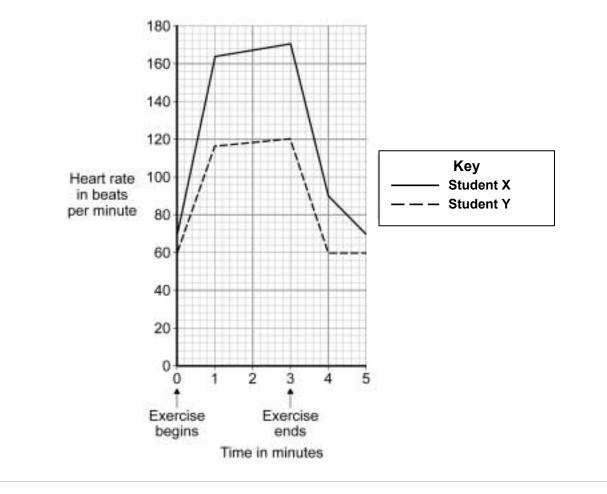
Question 13 continues on the next page

13 (b) Two students did the same step-up exercise for 3 minutes.

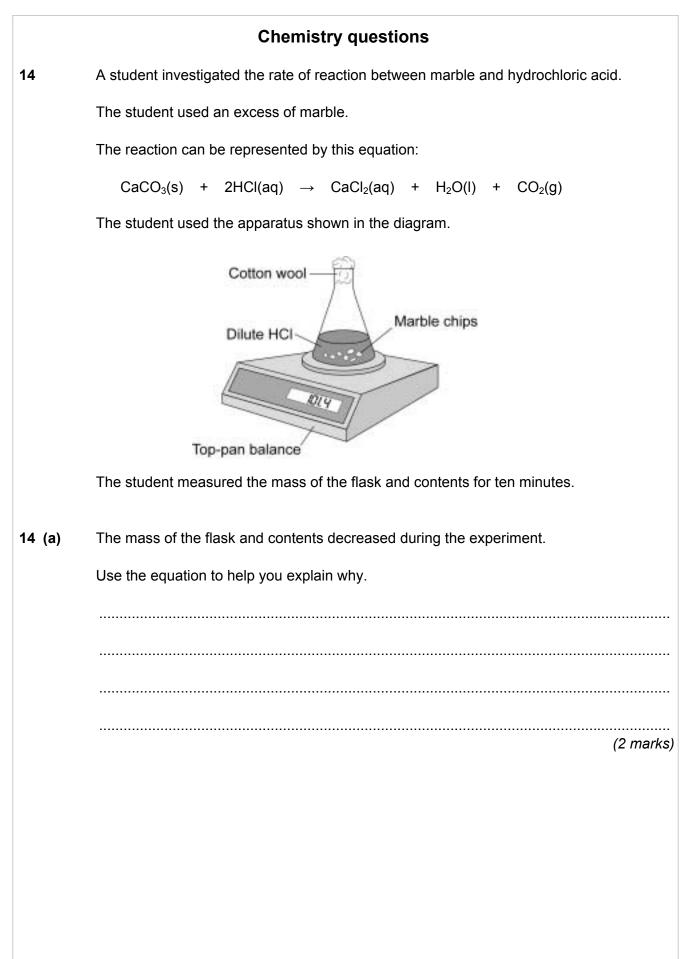


One of the students was fit. The other student was unfit.

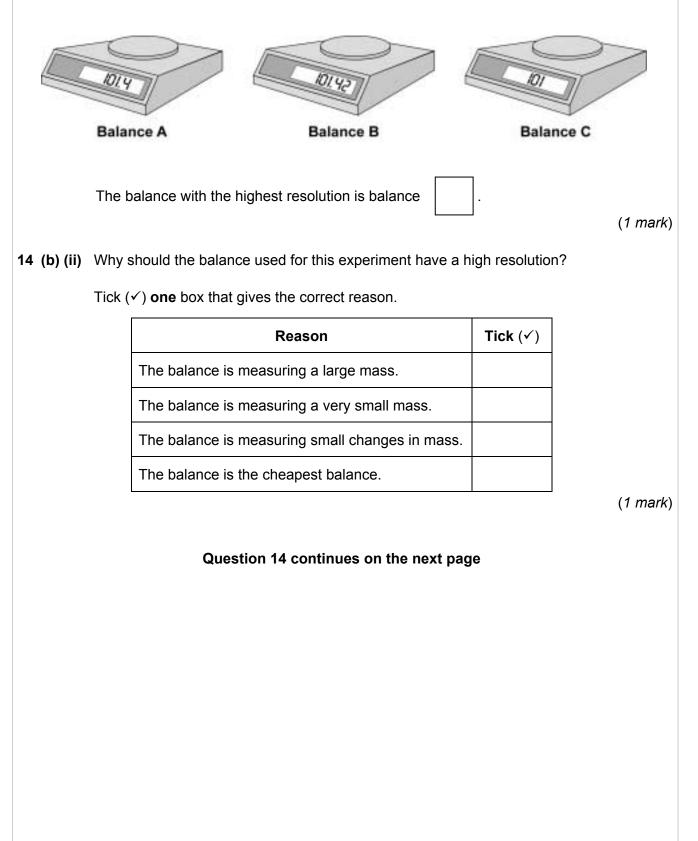
The graph shows how the students' heart rate changed during the exercise and after the exercise.

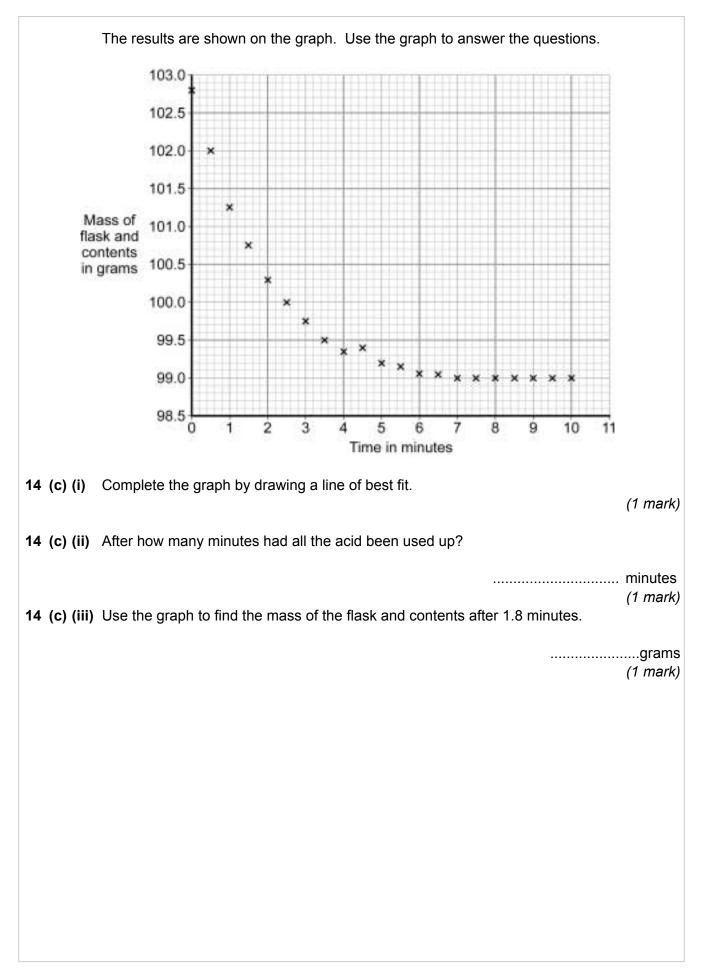


13 (b)	Suggest which student was the fitter.
	Draw a ring around the correct answer. Student X / Student Y
	Give three reasons for your answer.
	1
	2
	3
	(3 marks)
13 (c)	Explain the advantage to the students of the change in heart rate during exercise.
	(4
	(4 marks)



- **14 (b)** The balance used to measure the mass of the apparatus must be of high resolution for this experiment.
- 14 (b) (i) Which balance, A, B, or C, has the highest resolution?





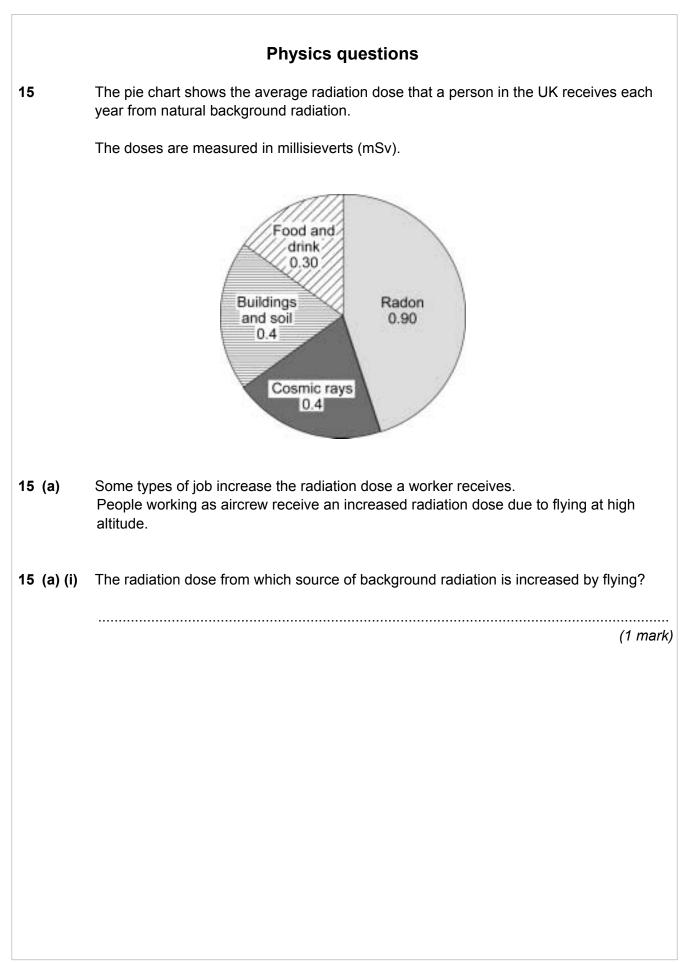
14 (d) The student repeated the experiment using powdered marble instead of lumps of marble.

The rate of reaction between the marble and hydrochloric acid particles was much faster with the powder.

Explain why.

(2 marks)

Turn over for the next question



15 (a) (ii) The following table gives the average additional radiation dose received by aircrew flying to various destinations from London.

Destination	Flight time in hours	Average additional radiation dose in mSv
Edinburgh	1	0.004
Istanbul	5	0.025
Toronto	8	0.050
Los Angeles	11	0.065
Tokyo	13	0.075

What is the relationship between flight time and average additional radiation dose?

(1 mark)

15 (a) (iii) A flight from London to Jamaica takes 10 hours.

Estimate the likely value for the average additional radiation dose received by people on this flight.

Average additional radiation dose = mSv

Give a reason for your answer.

.....

.....

(2 marks)

Question 15 continues on the next page

15 (b) The following table gives the effects of different radiation doses on the human body. Radiation Effects dose in mSv 10000 Immediate illness; death within a few weeks 1000 Radiation sickness; unlikely to cause death 100 Lowest dose with evidence of causing cancer A businessman makes 10 return flights a year from London to Tokyo. Explain whether the businessman should be concerned about the additional radiation dose received during the flights. (2 marks) 15 (c) In a study of 3900 aircrew it was found that 169 had developed leukaemia, a form of cancer. In a similar sized sample of non-aircrew the number of leukaemia cases was 156. Suggest why it would be difficult to be certain that the leukaemia developed by the aircrew was caused by flying. (2 marks) END OF QUESTIONS Copyright © 2011 AQA and its licensors. All rights reserved.



GCSE Physics Equations Sheet

Unit 6 F and H

F	F resultant force
$a = \frac{1}{m}$ or $F = m \times a$	<i>m</i> mass
m	a acceleration
	a acceleration
v - u	v final velocity
$a = -\frac{1}{\iota}$	<i>u</i> initial velocity
	t time taken
	<i>W</i> weight
$W = m \times g$	<i>m</i> mass
	<i>g</i> gravitational field strength
	g gravitational new strength
	F force
$F - k \times e$	k spring constant
	e extension
	W work done
$W = F \times d$	F force applied
	<i>d</i> distance moved in the direction of the force
R	P power
$P = \frac{2}{t}$	E energy transferred
	<i>t</i> time taken
	E_{ρ} change in gravitational potential energy
E - m V a V b	<i>m</i> mass
$E_p = m \times g \times h$	g gravitational field strength
	<i>h</i> change in height

$E_k = \frac{1}{2} \times m \times v^2$	 <i>E_k</i> kinetic energy <i>m</i> mass <i>v</i> speed
$p = m \times v$	 <i>p</i> momentum <i>m</i> mass <i>v</i> velocity
$I = \frac{Q}{t}$	 <i>I</i> current <i>Q</i> charge <i>t</i> time
$V = \frac{W}{Q}$	V potential differenceW work doneQ charge
$V = I \times R$	 <i>V</i> potential difference <i>I</i> current <i>R</i> resistance
$P = \frac{E}{t}$	<pre>P power E energy t time</pre>
$P = I \times V$	 <i>P</i> power <i>I</i> current <i>V</i> potential difference
$E = V \times Q$	<i>E</i> energy<i>V</i> potential difference<i>Q</i> charge